ಎಸ್ಎಸ್ಎಲ್ಸಿ-ಇಂಗ್ಲಿಷ್ ಮಾಧ್ಯಮ

## **SCIENCE**

## **Carbon and its Compounds**

## VERY SHORT ANSWER TYPE OUESTIONS

4. Name the functional group present in each of the following organic compounds:

(i) CH<sub>3</sub>COCH<sub>3</sub>

(ii) C<sub>2</sub>H<sub>5</sub>COOH

Ans:

- (ii) Carboxylic acid (-COOH)
- 5. Write the name and formula of the second member of carbon compounds having functional group -COOH.

Ans:

6. Write the name and formula of the first member of the carbon compounds having functional group -COOH.

Ans:

Methanoic acid

7. Write the name and formula of the first member of the carbon compounds having functional group –CHO.

Ans:

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Methans

8. Write the name and formula of the  $2^{nd}$  member of the series of carbon compounds whose general formula is  $C_nH_{2n+1}$  OH.

Ans: Ethanol, C<sub>2</sub>H<sub>5</sub>OH or CH<sub>3</sub>CH<sub>2</sub>OH

9. Write the name and formula of the  $2^{nd}$  member of the series of carbon compounds whose general formula is  $C_nH_{2n}$ .

**Ans:**  $C_3H_6$ ,  $H_2C = CH - CH_3$ 

Propene is second member of series whose general formula is  $C_nH_{2n}$ .

10. How many covalent bonds are there in a molecule of ethane  $(C_2H_6)$ ?

**Ans:** There are 7 covalent bonds in a molecule of ethane.

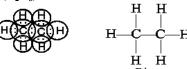
11. Write the electron dot structure of ethane molecule  $(C_2H_4)$ .

Ans:



12. Write the electron dot structure of ethane molecule ( $C_2H_6$ ).

Ans:



13. Draw the structure for ethanoic acid molecule, CH<sub>3</sub>COOH.

Ans:

Ethanoic acid

14. Draw the structure of butanone molecule, CH<sub>3</sub>COC<sub>2</sub>H<sub>5</sub>.

Ans:

15. Draw the structure of the hexanal molecule,  $C_5H_{11}\mathrm{CHO}$ .

Ans:

16. Butanone is a four carbon per molecule compound. Name the functional group present in it.

Ans: Ketone

17. Name the following compound:

Ans: 1-Hexyne is IUPAC name of the compound.

18. State two characteristic features of carbon which when put together give rise to large number of carbon compounds.

Ans: (i) Catenation

(ii) Tetravalency of carbon

19. Write the structural formula of chloroethane.
Ans:

20. Name the carbon compound which on heating with excess of concentrated sulphuric acid at 443K gives ethane.

Ans: CH<sub>3</sub>CH<sub>2</sub>OH, ethanol

$$CH_{3}CH_{2}OH \xrightarrow{\text{conc. } H_{2}SO_{4} \atop \text{443 K}} CH_{2} = CH_{2}$$
Ethanol